

Effect of mutations of *Thermoanaerobacter ethanolicus* secondary alcohol dehydrogenase at tryptophan-110 on enantioselectivity of reduction of phenyl-ring containing ketones



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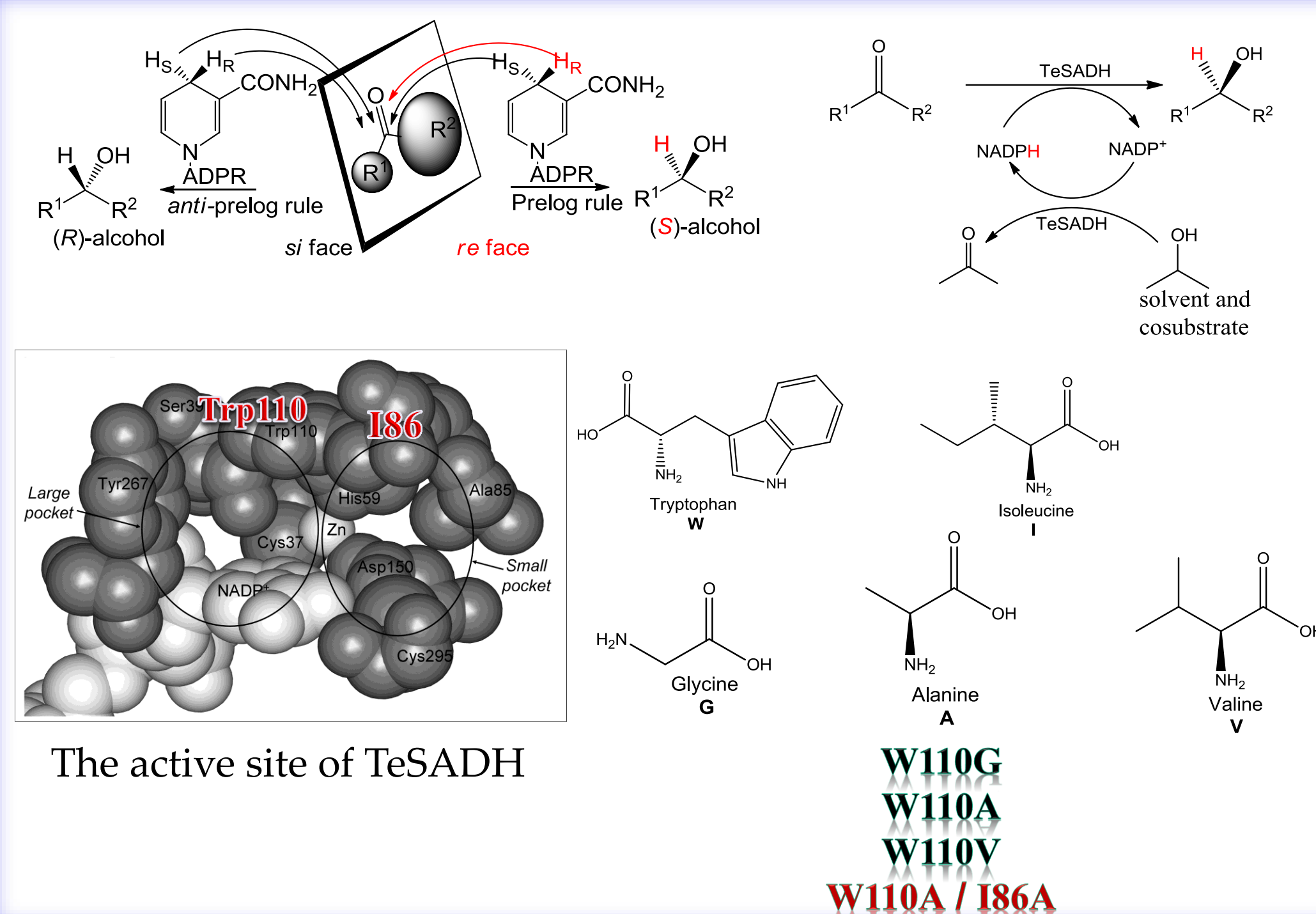
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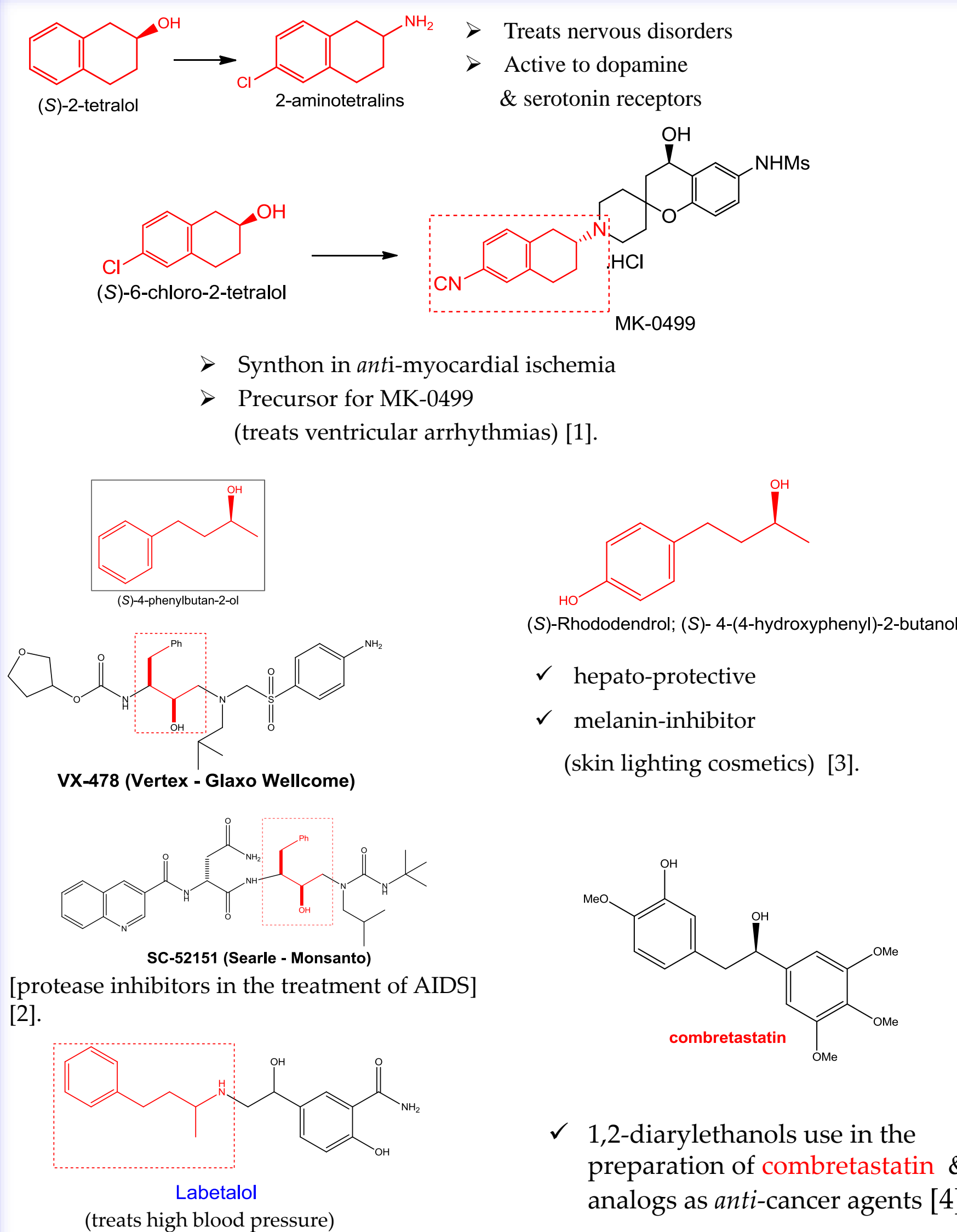
Abstract

- The asymmetric reduction of selected phenyl-ring-containing ketones by various mutants of *Thermoanaerobacter ethanolicus* secondary alcohol dehydrogenase (TeSADH) was studied using single and dual site mutagenesis.
- For substituted 2-tetralones we noticed that the changing in the position of the substituent on the aromatic ring also has a great impact on the binding affinity and maximum catalytic rate, as reflected by the kinetic parameters V_{max} and K_m .
- We also explored the regio- and enantioselective reduction of diketones using W110A/I86A TeSADH.
- The expansion of both small and large pockets of TeSADH in the mutant W110A/I86A not only accommodates the substrates of single mutants W110A and I86A within the active site, but also expands the substrate scope to ketones bearing two sterically demanding groups (bulky-bulky ketones); which are not substrates for TeSADH single mutants.

Introduction



Significance of Study



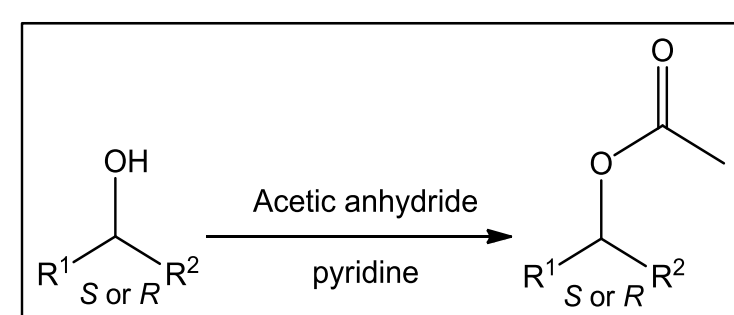
Materials and Methods

procedure for reaction

- Substrate (0.0191 mmol)
- TeSADH (W110G, W110A, W110V, W110A/I86A)
- Tris-HCl buffer solution (50 mM, pH 8.0)
- NADPH⁺ (1 mg)
- 50 °C, 24 hr.
- extraction with ethyl acetate

stereochemical outcome

- Polimetry : to measure the optical rotation of alcohol product
- GC equipped with a chiral column: to determine the *ee* of the corresponding acetate esters
- HPLC equipped with a chiral column

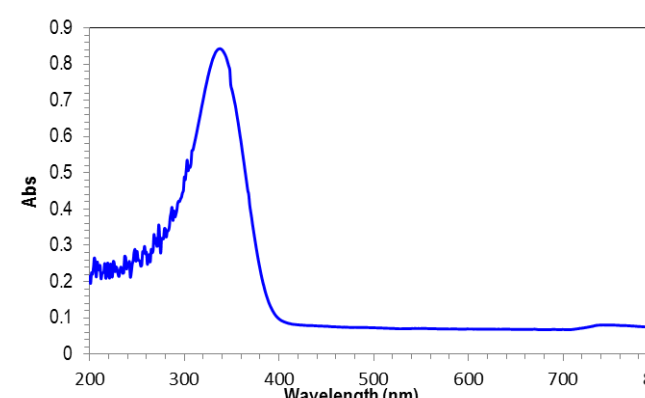


Chemical identity

- ¹H and ¹³C NMR
- IR
- EI MS

Enzyme kinetics

- Substrate concentration (0.001-10 mM)
- Tris-HCl buffer solutions (pH 6.5, 50 mM)
- CH₃CN (10%)
- NADPH (0.25 mM)
- 340 nm
- 50 °C



Results and Discussion

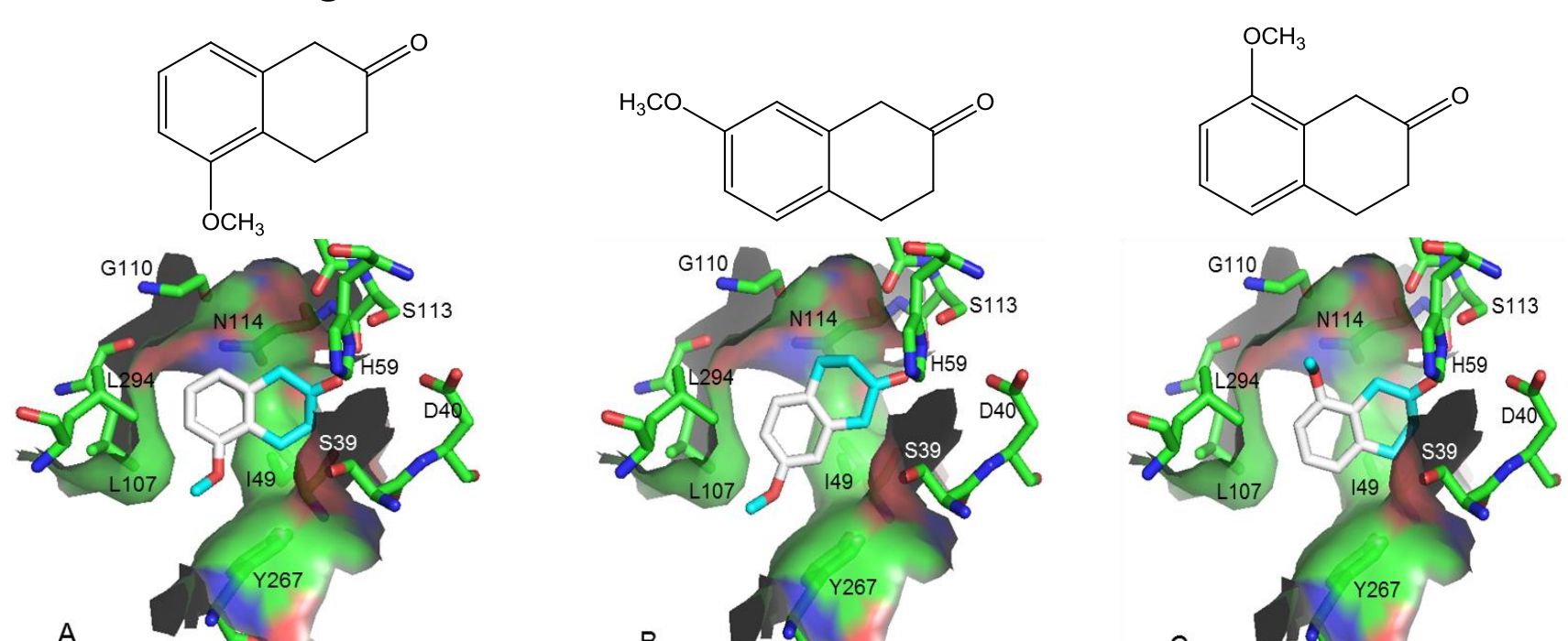
The asymmetric reduction of substituted 2-tetralones

Substrate	Product	Mutant TeSADH	Conv. (%)	ee (%)
W110G			>99	69
W110A			>99	71
W110V			5	91
W110A/I86A			12	87
W110G			>99	61
W110A			99	87
W110V			96	71
W110A/I86A			>99	95
W110G			>99	>99
W110A			99	>99
W110V			46	>99
W110A/I86A			>99	>99
W110G			>99	>99
W110A			99	>99
W110V			67	>99
W110A/I86A			94	>99
W110G			>99	>99
W110A			99	>99
W110V			86	>99
W110A/I86A			95	>99

Kinetics parameters for substituted 2-tetralones

Ketone	W110G TeSADH		W110A TeSADH	
	K_m (mM)	V_{max} (mM.min ⁻¹)	K_m (mM)	V_{max} (mM.min ⁻¹)
1: R ¹ =H, R ² =H 2: R ¹ =OCH ₃ , R ² =H 3: R ¹ =H, R ² =OCH ₃	0.0184 ± 0.000824	0.0748 ± 0.00335	0.00580 ± 0.00029	0.0589 ± 0.0048
1: R ¹ =H, R ² =H 2: R ¹ =OCH ₃ , R ² =H 3: R ¹ =H, R ² =OCH ₃	0.0231 ± 0.0011	0.0666 ± 0.00316	0.0106 ± 0.000528	0.0695 ± 0.0039
1: R ¹ =H, R ² =H 2: R ¹ =OCH ₃ , R ² =H 3: R ¹ =H, R ² =OCH ₃	0.0452 ± 0.00156	0.0743 ± 0.00257	0.0122 ± 0.000611	0.0545 ± 0.00218

Docking results for substituted 2-tetralones



Lowest energy dockings of A:5-methoxy-2-tetralone, B:7-methoxy-2-tetralone, C:8-methoxy-2-tetralone in the catalytic site of W110G TeSADH. The substrate is viewed from the face occupied by NADP in the TeSADH crystal structure.

The asymmetric reduction of substituted 4-phenyl-2-butanone

Substrate	Product	Mutant TeSADH	Conv. (%)	ee (%)
W110G			>99	96
W110A			99	99
W110V			95	>99
W110A/I86A			91	98
W110G			93	20
W110A			98	60
W110V			94	79
W110A/I86A			40	84
W110G			95	99
W110A			97	>99
W110V			92	94
W110A/I86A			80	>99

The asymmetric reduction of substituted phenyl acetone

Substrate	Configuration of Product	Mutant TeSADH	Conv. (%)	ee (%)
(S)	(S)	W110G	>99	79
(S)	(S)	W110A	95	84
(S)	(S)	W110V	>99	>99
(R)	(R)	W110A/I86A	>99	80
(S)	(S)	W110G	13	89
(S)	(S)	W110A	84	98
(S)	(S)	W110V	99	>99
(R)	(R)	W110A/I86A	>99	99

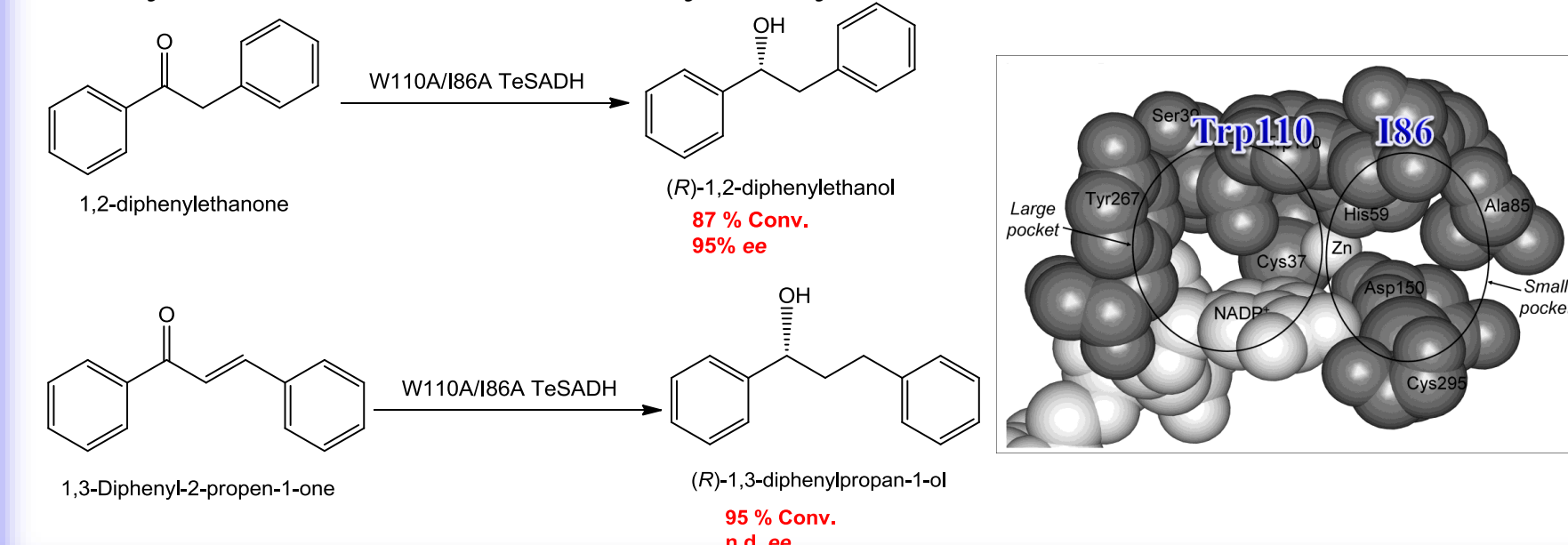
The asymmetric reduction of other aromatic ketones

Substrate	Products	Mutant TeSADH	Conv. (%)	ee (%)
1-phenylpropan-2-one	(S)-1-phenylpropan-2-ol	W110G	36	93
		W110A	98	92
		W110V	98	>99
		W110A/I86A	72	71
1-phenylbutan-2-one	(R)-1-phenylbutan-2-ol	W110G	nr	n.d.
		W110A	3	n.d.
		W110V	32	92
		W110A/I86A	95	98
1-phenylpentan-2-one	(R)-1-phenylpentan-2-ol	W110G	nr	nr
		W110A	nr	nr
		W110V	79	79
		I86A	79	79
		W110A/I86A	46	46

Regioselective reduction of aromatic ketones

Substrate	Product	Mutant TeSADH	Conv. (%)	ee (%)
1-phenylbutane-1,3-dione	(S)-3-hydroxy-1-phenylbutane-1-one	W110G	92	>99
		W110A	99	>99
		W110V	89	>99
		W110A/I86A	92	92
1-phenylpropane-1,2-dione	(R)-2-hydroxy-1-phenylpropane-1-one	W110G	3	n.d.
		W110A	8	n.d.
		W110V	>99	>99
		W110A/I86A	2	n.d.
		W110G	nr	nr
		W110A	nr	nr
		W110V	nr	nr
		W110A/I86A	>99	>99

Asymmetric reduction of bulky-bulky aromatic ketones



Conclusion

- W110A/I86A TeSADH expanded the substrate scope for TeSADH.
- Various mutants of TeSADH reduced diketones in high regioselectivity and enantioselectivity.
- A reversed enantiopreference for few substrates was observed when W110A/I86A TeSADH was used as a catalyst.
- The interactions between the ketone and the residues of the active site of enzyme determine the mode of substrate binding, and thus the enzyme's stereoselectivity.

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